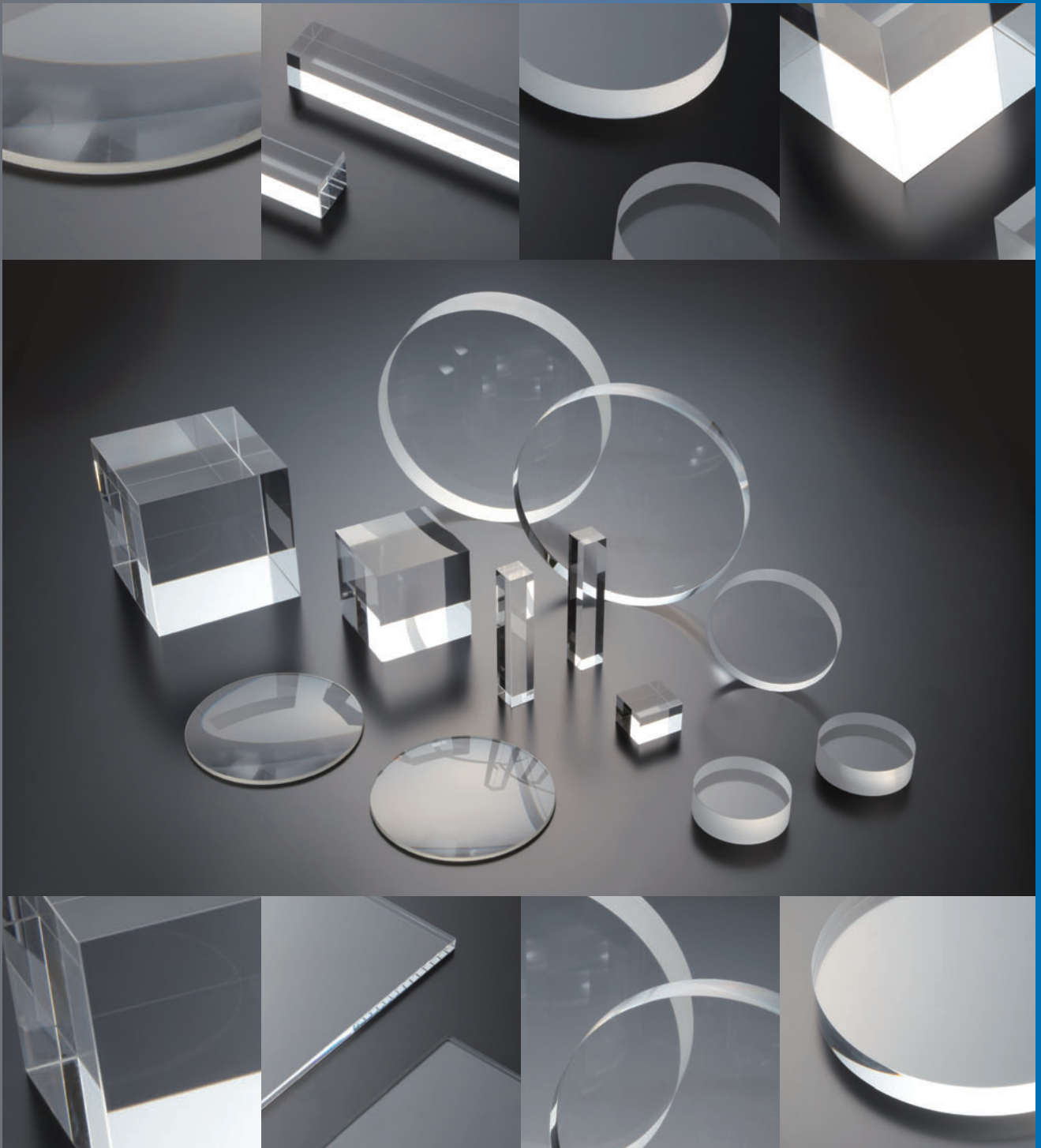
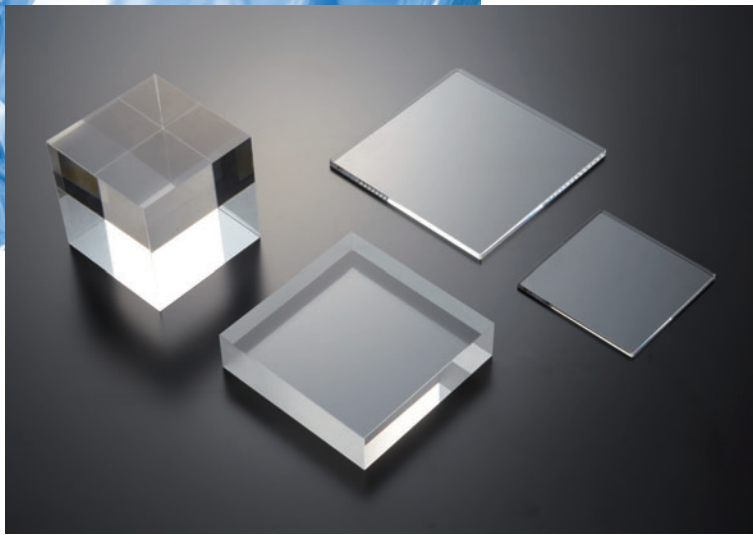


Optical Silica Glass



ES Series



ES ingots are manufactured from high-purity SiCl₄ using an oxy-hydrogen flame in a verneuil type process.

For very large parts, such as LCD photomasks, ES ingots are reflowed into large rectangular blocks.

ES ingots have a typical purity that exceeds the detection limit of 1 Oppb in all elements, and are virtually free of bubbles and inclusions.

As a result, ES is the material of choice for LCD photomasks up to Gen. 10, and selected subgrades are widely used in leading edge stepper (KrF and ArF) and optics applications.

Grade

Grades	Features	Application	Available size(sidexside. Diameter)
ES	Standard optical grade for planar optics	LCD Mask, Mirror&window substrate	1,800x1,700mm *1
ESL-1	Standard optical grade with 1D striae free	General Optics & laser Optics	1,300x1,000mm *1
ESL-1000	Enhance excimer durability for ESL-1	Stepper lens (illumination)	1,300x1,000mm *1
ESL-2	Good homogeneity with 3D striae free	High precision optics	~Φ450mm
ESL-2000	Enhance excimer durability for ESL-2	Stepper lens (Projection)	~Φ450mm

*1 Reflow ingots

Optical Properties

Grades	Striae *2	Homogeneity *3 Δn(x10 ⁻⁹) within CA	Strain (nm/cm)	Fluorescence *4	Excimer Durability	Applicable Wavelength *6 (nm)	(Typical values)
							Bubble Class *7
ES	n.sp.	n.sp.	1~20	n.sp.	n.sp.	180~2100	Class 0
ESL-1	1D	5~15(CA:Φ500mm) 3~10(CA:Φ300mm)	1~10	None	n.sp.	180~2100	Class 0
ESL-1000	1D	5~15(CA:Φ500mm) 3~10(CA:Φ300mm)	1~10	None	sp. *5	180~2100	Class 0
ESL-2	3D	<10(CA:Φ300mm)	1~10	None	n.sp.	180~2100	Class 0
ESL-2000	3D	<10(CA:Φ300mm)	1~10	None	sp. *5	180~2100	Class 0

*2 Striae specification is defined as below and confirmed by interferometric measurement.
1D : 1 direction free 3D : 3 direction free

n.sp. = not specified

*3 Homogeneity means refractive index homogeneity and is expressed as Δn(PV) within effective area.

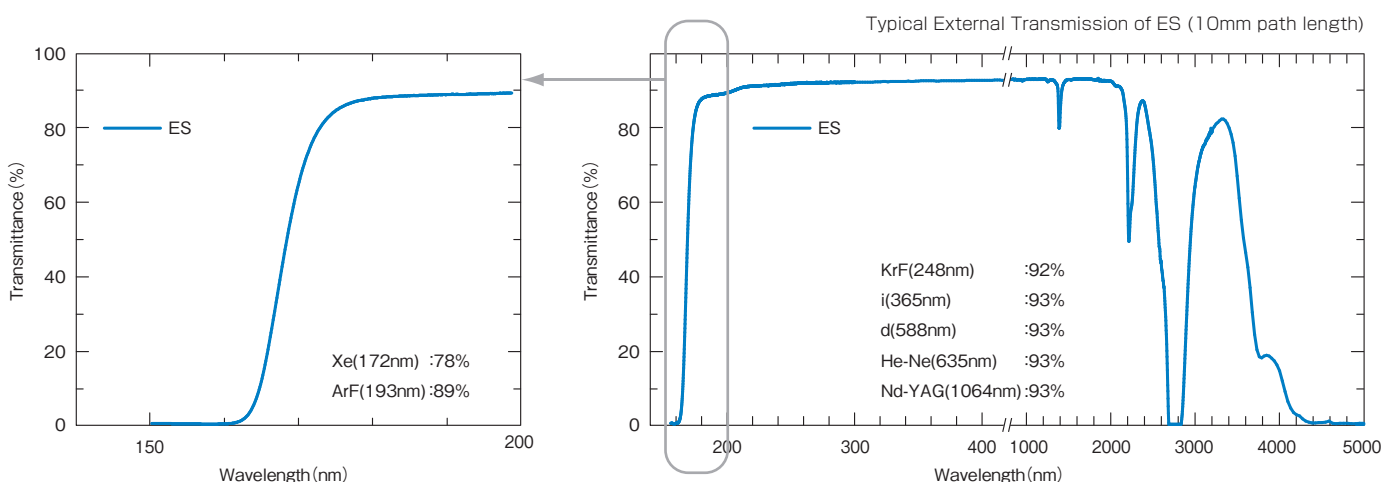
*4 Fluorescence is checked with excitation at 254nm using an Hg-lamp.

*5 Excimer durability is specified with Tosoh's KrF and ArF excimer laser damage tests.

*6 Applicable wavelength is defined as the wavelength range where external transmittance is more than 80%/cm.

*7 Bubble class is as per DIN 58927.

Spectral Transmission



Optical Properties

λ (μm)		Refractive Index ^{*1}
n_{1530}	1.53000	1.44428
n_{1064}	1.06400	1.44965
n_t	1.01398	1.45027
n_s	0.85211	1.45249
$n_{A'}$	0.76819	1.45392
n_{708}	0.70820	1.45514
n_{691}	0.69075	1.45554
n_{672}	0.67164	1.45600
n_c	0.65627	1.45640
$n_{c'}$	0.64385	1.45674
$n_{\text{He-Ne}}$	0.63299	1.45704
n_{623}	0.62344	1.45733
n_D	0.58929	1.45844
n_d	0.58773	1.45849
n_{577}	0.57696	1.45888
n_e	0.54623	1.46011
n_{492}	0.49161	1.46284
n_F	0.48613	1.46317
$n_{F'}$	0.47999	1.46354
n_g	0.43596	1.46673
n_h	0.40477	1.46966
n_i	0.36512	1.47458
n_{334}	0.33415	1.47983
n_{313}	0.31266	1.48454
n_{289}	0.28936	1.49107
n_{270}	0.26988	1.49814
n_{254}	0.25373	1.50557
n_{248}	0.24800	1.50865
n_{229}	0.22887	1.52121
n_{215}	0.21451	1.53378
n_{194}	0.19423	1.55899
n_{185}	0.18495	1.57511

Refractive Index		Abbe No.		Dispersion	
n_d	1.45849	v_d	67.72	n_F-n_C	0.00677
n_e	1.46011	v_e	67.58	$n_{F'}-n_{C'}$	0.00681

Partial Dispersion		Partial Dispersion Ratio	
n_c-n_t	0.00613	$\theta_{c,t}$	0.9056
$n_{c'}-n_{A'}$	0.00248	$\theta_{c,A'}$	0.3665
n_d-n_c	0.00209	$\theta_{d,c}$	0.3093
n_e-n_c	0.00371	$\theta_{e,c}$	0.5480
n_g-n_d	0.00824	$\theta_{g,d}$	1.2167
n_g-n_F	0.00356	$\theta_{g,F}$	0.5260
n_h-n_g	0.00293	$\theta_{h,g}$	0.4323
n_i-n_g	0.00785	$\theta_{i,g}$	1.1600
n_c-n_t	0.00647	$\theta'_{c,t}$	0.9499
n_e-n_c	0.00337	$\theta'_{e,c}$	0.4952
$n_{F'}-n_e$	0.00344	$\theta'_{F',e}$	0.5046
$n_i-n_{F'}$	0.01104	$\theta'_{i,F'}$	1.6209

Sellmeier Dispersion Formula for Refractive Index			
$n^2 - 1 = A_1 \lambda^2 / (\lambda^2 - B_1^2) + A_2 \lambda^2 / (\lambda^2 - B_2^2) + A_3 \lambda^2 / (\lambda^2 - B_3^2)$			
Constants of Sellmeier Dispersion Formula			
A1	0.6961852	B1	0.0685886
A2	0.4079779	B2	0.1162043
A3	0.8974798	B3	9.8961609

Differential Temperature Coefficients of the Refractive Index								
Temp. range (°C)	dn/dT absolute ($\times 10^{-6}/\text{°C}$)				dn/dT relative ($\times 10^{-6}/\text{°C}$)			
	e (546.1nm)	g (435.8nm)	h (404.7nm)	i (365.0nm)	e (546.1nm)	g (435.8nm)	h (404.7nm)	i (365.0nm)
-40~-20	7.8	8.2	8.4	8.7	9.8	10.2	10.5	10.8
-20~0	8.2	8.6	8.9	9.2	9.9	10.4	10.7	11.0
0-20	8.7	9.1	9.4	9.7	10.1	10.6	10.9	11.2
20~40	9.1	9.5	9.8	10.2	10.6	10.8	11.1	11.5
40~60	9.5	9.9	10.2	10.6	10.6	11.1	11.4	11.7
60~80	9.8	10.3	10.5	10.9	10.8	11.3	11.6	12.0

Internal Transmission	
λ (nm)	$t=10\text{mm}$
170	77
175	95
180	98
190	98
200	99
210	100
220	100
230	100
240	100
250	100
260	100
270	100
280	100
290	100
300	100
350	100
400	100
450	100
500	100
600	100
700	100
800	100
900	100
1000	100
1200	100
1400	93
1600	100
1800	100
2000	99
2100	98
2200	66

*1 All refractive indices are calculated from values measured under dry N2 at 25°C, 1013hPa

Mechanical Properties

Item	Value
Density (g/cm^3)	2.2
Young's modulus (GPa)	74
Shear modulus (GPa)	31
Poisson's ratio	0.18
Bending strength ^{*2} (MPa)	65~95
Compressive strength (MPa)	1,130
Tensile strength ^{*2} (MPa)	49
Torsion strength (MPa)	29
Vickers hardness (MPa)	8,900

Thermal Properties

Item	Value
Strain point ($\eta=10^{14.5}$) (°C)	980
Annealing Point ($\eta=10^{13}$) (°C)	1,080
Softening Point ($\eta=10^{7.6}$) ^{*3} (°C)	(1,720)
Thermal Conductivity (W/mK) at 20°C	1.38
Specific heat (J/kg·K) at 20°C	749
Thermal diffusivity ($\times 10^{-7}\text{m}^2/\text{s}$) at 20°C	8.5
Coefficient of thermal expansion ($\times 10^{-7}/\text{°C}$)	
$\alpha: 30\text{°C} \sim 100\text{°C}$	5.2
$\alpha: 30\text{°C} \sim 200\text{°C}$	5.8
Viscosity log η (Poise) at 1,200°C	11.4

Electrical Properties

Item	Value
Dielectric constant at 500MHz	3.9
Dielectric loss factor ($\tan \delta$) at 500MHz	$< 1 \times 10^{-3}$
Resistivity (Ω)	5×10^{15}
Volume resistivity ($\Omega \cdot \text{cm}$)	1×10^{17}

*2 Bending and Tensile strengths are affected by surface conditions.

*3 Estimate from extrapolation

NOTE : Unless otherwise stated, all values represent typical data at 25°C

ED-H



ED-H ingots are manufactured from high-purity SiCl_4 in a VAD process.

ED-H ingots have a typical purity that exceeds the detection limit of 10ppb in all elements, and are virtually free of bubbles and inclusions.

Thanks to an OH content less than 100ppm, ED-H material exhibits superior transmission characteristics in the deep UV and is a material of choice for VUV applications down to 170nm.

Grade

Grade	Feature	Application	Available size (Diameter)
ED-H	Good homogeneity with 3D striae free	High precision optics	Φ180mm

Optical Properties

(Typical values)

Grade	Striae *1	Homogeneity *2 $\Delta n(\times 10^{-9})$ within CA	Strain (nm/cm)	Fluorescence *3	Excimer Durability	Applicable Wavelength *4 (nm)	Bubble Class *5
ED-H	3D	5~10 (CA : Φ180mm)	1~10	None	n.sp.	170~2,600	Class 0

*1 Striae specification is defined as below and confirmed by interferometric measurement.
1D : 1 direction free 3D : 3 direction free

n.sp. = not specified

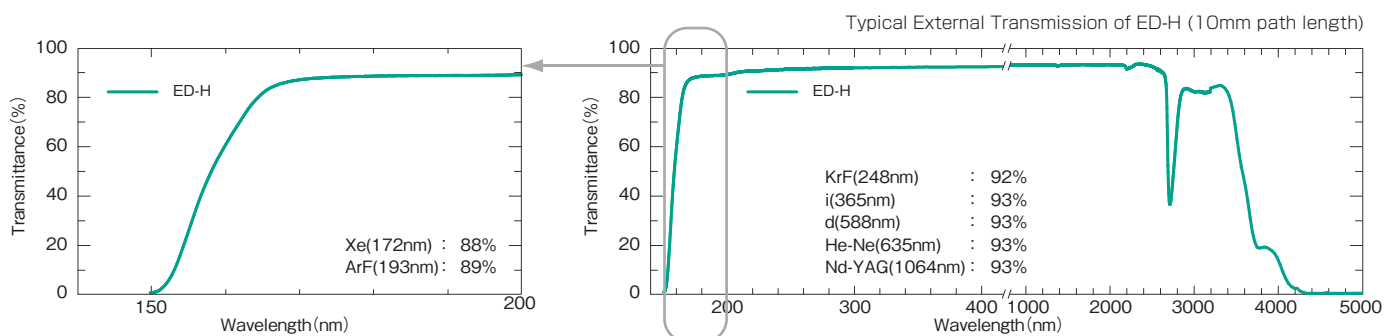
*2 Homogeneity means refractive index homogeneity and is expressed as $\Delta n(\text{PV})$ within effective area.

*3 Fluorescence is checked with excitation at 254nm using an Hg-lamp.

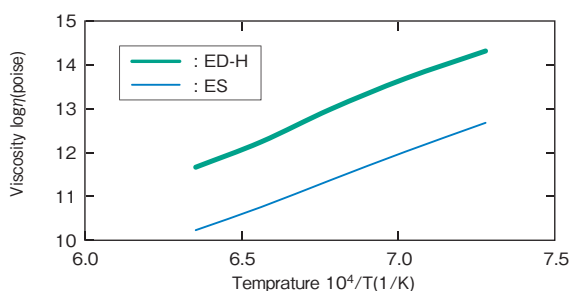
*4 Applicable wavelength is defined as the wavelength range where external transmittance is more than 80%/cm.

*5 Bubble class is as per DIN 58927.

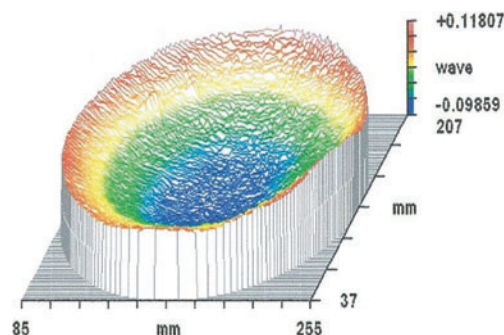
Spectral Transmission



High temperature viscosity



Homogeneity (Typical mapping data)



Optical Properties

λ (μm)		Refractive Index ^{*1}	Refractive Index		Abbe No.		Dispersion		Internal Transmission			
n_{1530}	1.53000	1.44440	n_d	1.45861	v_d	67.75	n_F-n_C	0.00677	$\lambda(\text{nm})$	$t=10\text{mm}$		
n_{1064}	1.06400	1.44977	n_e	1.46022	v_e	67.61	n_F-n_C	0.00681	170	98		
n_t	1.01398	1.45038	Partial Dispersion		Partial Dispersion Ratio				175	99		
n_s	0.85211	1.45260	n_C-n_t	0.00613	$\theta_{C,t}$	0.9056			180	99		
$n_{A'}$	0.76819	1.45403	$n_C-n_{A'}$	0.00248	$\theta_{C,A'}$	0.3665			190	99		
n_{708}	0.70820	1.45525	n_d-n_C	0.00209	$\theta_{d,C}$	0.3094			200	99		
n_{691}	0.69075	1.45565	n_e-n_C	0.00371	$\theta_{e,C}$	0.5481			210	99		
n_{672}	0.67164	1.45611	n_g-n_d	0.00823	$\theta_{g,d}$	1.2166			220	100		
n_C	0.65627	1.45651	n_g-n_F	0.00356	$\theta_{g,F}$	0.5259			230	100		
$n_{C'}$	0.64385	1.45685	n_h-n_g	0.00293	$\theta_{h,g}$	0.4324			240	100		
n_{He-Ne}	0.63299	1.45716	n_i-n_g	0.00785	$\theta_{i,g}$	1.1600			250	100		
n_{623}	0.62344	1.45744	n_C-n_t	0.00647	$\theta'_{C,t}$	0.9502			260	100		
n_D	0.58929	1.45855	n_e-n_C	0.00337	$\theta'_{e,C}$	0.4954			270	100		
n_d	0.58773	1.45861	n_F-n_e	0.00344	$\theta'_{F,e}$	0.5046			280	100		
n_{577}	0.57696	1.45899	n_i-n_F	0.01104	$\theta'_{i,F}$	1.6213			290	100		
n_e	0.54623	1.46022	Sellmeier Dispersion Formula for Refractive Index								300	100
n_{492}	0.49161	1.46296	$n^2-1 = A_1 \lambda^2 / (\lambda^2 - B_1^2) + A_2 \lambda^2 / (\lambda^2 - B_2^2) + A_3 \lambda^2 / (\lambda^2 - B_3^2)$								350	100
n_F	0.48613	1.46328	Constants of Sellmeier Dispersion Formula								400	100
$n_{F'}$	0.47999	1.46366	A1	0.6963511	B1	0.0685337			450	100		
n_g	0.43596	1.46684	A2	0.4081467	B2	0.1162167			500	100		
n_h	0.40477	1.46977	A3	0.8974786	B3	9.8961611			600	100		
n_i	0.36512	1.47469							700	100		
n_{334}	0.33415	1.47994							800	100		
n_{313}	0.31266	1.48465							900	100		
n_{289}	0.28936	1.49118							1000	100		
n_{270}	0.26988	1.49825							1200	100		
n_{254}	0.25373	1.50568							1400	99		
n_{248}	0.24800	1.50876							1600	100		
n_{229}	0.22887	1.52132							1800	100		
n_{215}	0.21451	1.53389							2000	100		
n_{194}	0.19423	1.55910							2200	98		
n_{185}	0.18495	1.57522							2400	100		
									2600	97		
									2800	71		

*1 All refractive indices are calculated from values measured under dry N2 at 25°C, 1013hPa

Mechanical Properties^{*2}

Item	Value
Density (g/cm ³)	2.2
Young's modulus (GPa)	74
Shear modulus (GPa)	31
Poisson's ratio	0.18
Bending strength ^{*2} (MPa)	65~95
Compressive strength (MPa)	1,130
Tensile strength ^{*2} (MPa)	49
Torsion strength (MPa)	29
Vickers hardness (MPa)	8,900

Thermal Properties

Item	Value
Strain point ($\eta=10^{14.5}$) (°C)	1,090
Annealing Point ($\eta=10^{13}$) (°C)	1,190
Softening Point ($\eta=10^{7.6}$)*3 (°C)	(1,720)
Thermal Conductivity (W/mK) at 20°C	1.38
Specific heat (J/kg·K) at 20°C	749
Thermal diffusivity ($\times 10^{-7} \text{m}^2/\text{s}$) at 20°C	8.5
Coefficient of thermal expansion ($\times 10^{-7}/\text{°C}$)	
$\alpha: 30\text{°C} \sim 100\text{°C}$	5.2
$\alpha: 30\text{°C} \sim 200\text{°C}$	5.8
Viscosity log η (Poise) at 200°C	13.0

Electrical Properties

Item	Value
Dielectric constant at 500MHz	3.9
Dielectric loss factor ($\tan \delta$) at 500MHz	$< 1 \times 10^{-3}$
Resistivity (Ω)	8×10^{15}
Volume resistivity ($\Omega \cdot \text{cm}$)	5×10^{17}

*2 Bending and Tensile strengths are affected by surface conditions.

*3 Estimate from extrapolation

NOTE : Unless otherwise stated, all values represent typical data at 25°C

Impurity Level of Optical Silica Glass (Typical Value)

Unit(ppm)

Grades	Al	Ca	Cu	Fe	Na	K	Li	Mg	OH
ES	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	1000
ED-H	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<100

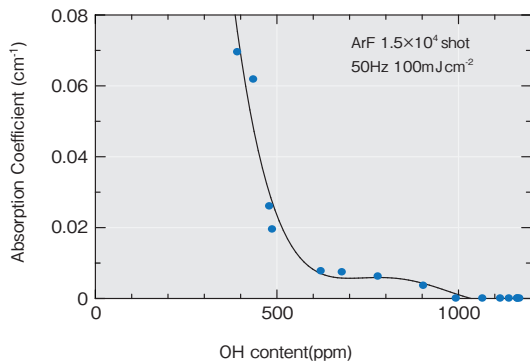
Fluorescence Property of ES series

ES Series do not show fluorescence when irradiating with 254nm light from a low pressure lamp. However, when irradiating with a strong light source such as X-ray and excimer lasers, fluorescence may occur. The following indicates the fluorescence seen in ES Series when irradiated with X-ray and UV lights.

Irradiated Light	Excitation Wavelength (nm)	Condition	Shot Number	Absorption Band	Visible Fluorescence
Low pressure Hg	254	Continuously, 2mW/cm ²	10,000	None	None
KrF excimer laser	248	25Hz, 200mJ/cm ²	10,000	None	None
ArF excimer laser	193	125Hz, 500mJ/cm ²	10,000	None	light red
ArF excimer laser	193	100Hz, 150mJ/cm ²	10,000	None	blue
ArF excimer laser	193	100Hz, 400mJ/cm ²	10,000	None	blue & slight red
X-ray	—	W, 50KV, ca 1E+6 rad	—	*1	—

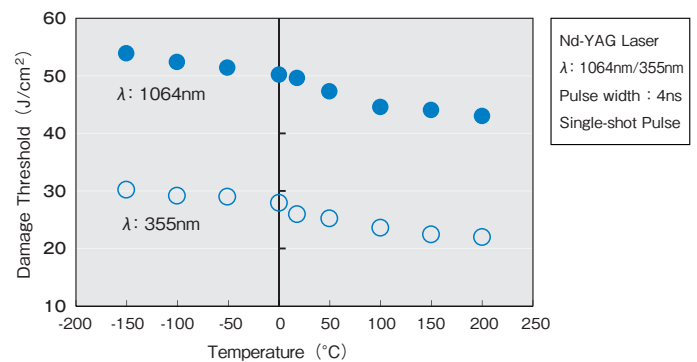
*1 Transmittance decrease less than 1% in the 200 to 300nm wavelength

Excimer Laser Durability of ES grades



Dependence of ArF Laser durability on OH Content

Nd: YAG laser damage threshold of ES



(K.Mikami, T.Jitsuno, et al., 56th JSAP conference 2009)

Chemical properties

Mass Loss of Clear Fused Quartz by Selected Acids

Solution	Conditions			Mass Loss (g/m ²)
	Conc. (%)	Temp. (°C)	Time (hr)	
H ₂ SO ₄	96	20	240	0.016
		205	24	0.06
HNO ₃	65	20	240	0.06
		115	24	0.11
HCl	37	20	240	0.18
		66	24	0.14

Mass Loss of Clear Fused Quartz by Selected Alkalis

Solution	Conditions			Mass Loss (g/m ²)
	Conc. (%)	Temp. (°C)	Time (hr)	
NH ₄ OH	10	25	72	0.06
		50	72	0.32
NaOH	5	25	96	0.135
		100	10	15
KOH	10	25	72	0.14
		100	10	11.3

Chemical reactivity towards other materials

Metals and nonmetals		Gases	
Al, Ag	Rapid reaction at 700-800°C	CO, SO ₂	No reaction
Au, Ag, Pt	No reaction	N ₂ , O ₂	No reaction
Zn, Sn, Pb	No reaction	Cl ₂	No reaction
Si	Slight reaction when fused	F ₂	No reaction with dried gases under 300°C
Ge	No reaction at 900°C	H ₂	No reaction
Mo, W	No reaction	HCl	No reaction
Oxides		Salts	
Al ₂ O ₃	Gradual reaction over 900°C	BaCl ₂	Reaction when fused
CaO	Reaction over 900°C	BaSO ₄	Reaction over 700°C
CuO	Reaction over 800°C	CaCl ₂	Slight reaction when fused at 800°C
Fe ₂ O ₃	Reaction over 900°C	KCl	Acceleration of devitrification at high temp.
PbO	Intense reaction with fusion	KF	Intense reaction when fused
MgO	Slight reaction at 900°C	NaCl	Reaction visually recognized over 800°C
ZnO	Reaction over 420°C	Na ₂ SO ₄	No reaction

Devitrification

When silica glass is exposed to high temperatures, the pure SiO₂ structure changes from a glass state (amorphous) to a stable crystalline state called cristobalite. This structural change is known as devitrification and generally occurs at temperatures over 1,150°C for clean clear fused quartz. Devitrification may also occur at temperatures below 1,000°C in the presence of impurities such as metal. The relation between the devitrification rate of clear fused quartz and temperature in various atmospheres is indicated below.

Gas composition	Temp.(°C)	Time(h)	Degree of devitrification	Devitrification thickness(μm)
Air	1,300	72	Surface completely devitrified	250
Dried oxygen	1,300	72	Devitrification of 50% of the surface	100~150
Industrial nitrogen	1,300	72	Surface devitrified	—
Nitrogen(O ₂ and H ₂ O removed)	1,300	72	No devitrification	—
Hydrogen(O ₂ and H ₂ O removed)	1,300	72	No devitrification	—

Handling Precautions

Care must be taken to avoid direct hand contact with silica glass. The skin's natural salts contain alkali such as sodium, potassium and other impurities that accelerate devitrification. All sources of metal contaminants should be avoided.

As a further precaution, fused silica should be washed in pure or distilled water, then either air dried in a clean area or wiped dry with an alcohol-soaked clean cloth. For more rigorous cleaning, a very thin surface layer of the glass can be removed by etching, prior to water washing, in a 5% - 10% hydrofluoric acid solution.

Usage Precautions

- * Always clean silica glass prior to use.
- * Dry product completely before using at high temperature.
- * Pay attention to devitrification due to atmospheric exposure.
- * Please refer to the thermal properties for your application. Fused silica can resist sudden heating and quenching, but it does have its limits.
- * Always consider fused silica's very low thermal expansion when the glass is used with other materials to avoid failure due to the differences in thermal expansion.
- * Take caution during prolonged usage at temperatures approaching the annealing point.
- * Be aware that slow sagging may occur under high temperature.



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